

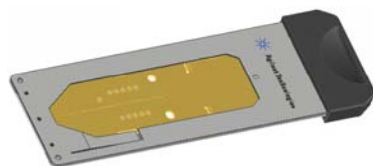
Tips and Tricks of HPLC Separation

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Agilent Technologies, Inc.
June 24, 2009



Goals for this presentation:

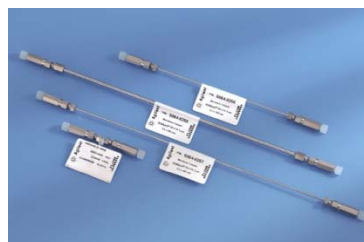
1. Introduce the most commonly observed column related problems in HPLC.
2. Explore the reasons for these column problems.
3. Propose preventative maintenance and method development/optimization approaches to minimize HPLC column problems and increase column lifetimes.



Chip LC



Nano LC



Capillary LC

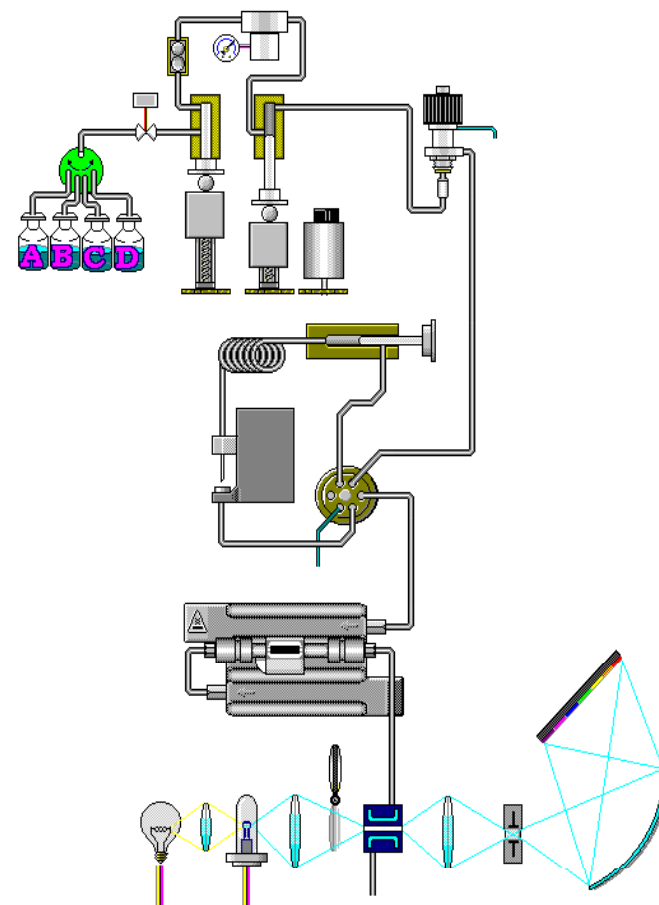
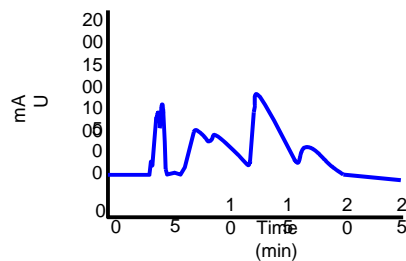
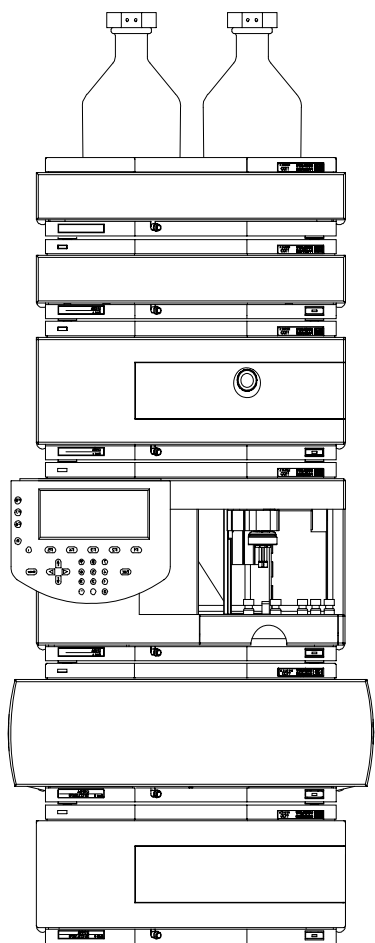


Analytical LC



Prep LC

Troubleshooting in HPLC



Major Areas of Column Problems - Dramatic Changes in 3 Key Areas:

- 1. HPLC System Pressure**
- 2. Chromatogram - Peak Shape**
- 3. Chromatogram - Peak Retention/Selectivity**

1. Pressure Issues

Column Observations

Large pressure change

Potential Problems

Plugged inlet frit

Column contamination

Plugged packing



Determining the Cause and Correcting High Back Pressure

- Check pressure with/without column - many pressure problems are due to blockages elsewhere in the system.

If Column pressure remains high:

- Rinse column (**remove detector from flow path!**)
 - Eliminate column contamination and plugged packing
 - high molecular weight/adsorbed compounds
 - precipitate from sample or buffer
- Back flush column – may clear plugged column inlet frit
- Change column inlet frit (... or discard column)

Eliminate pressure issues – add a disposable 0.5 or 2 um in-line filter to system.

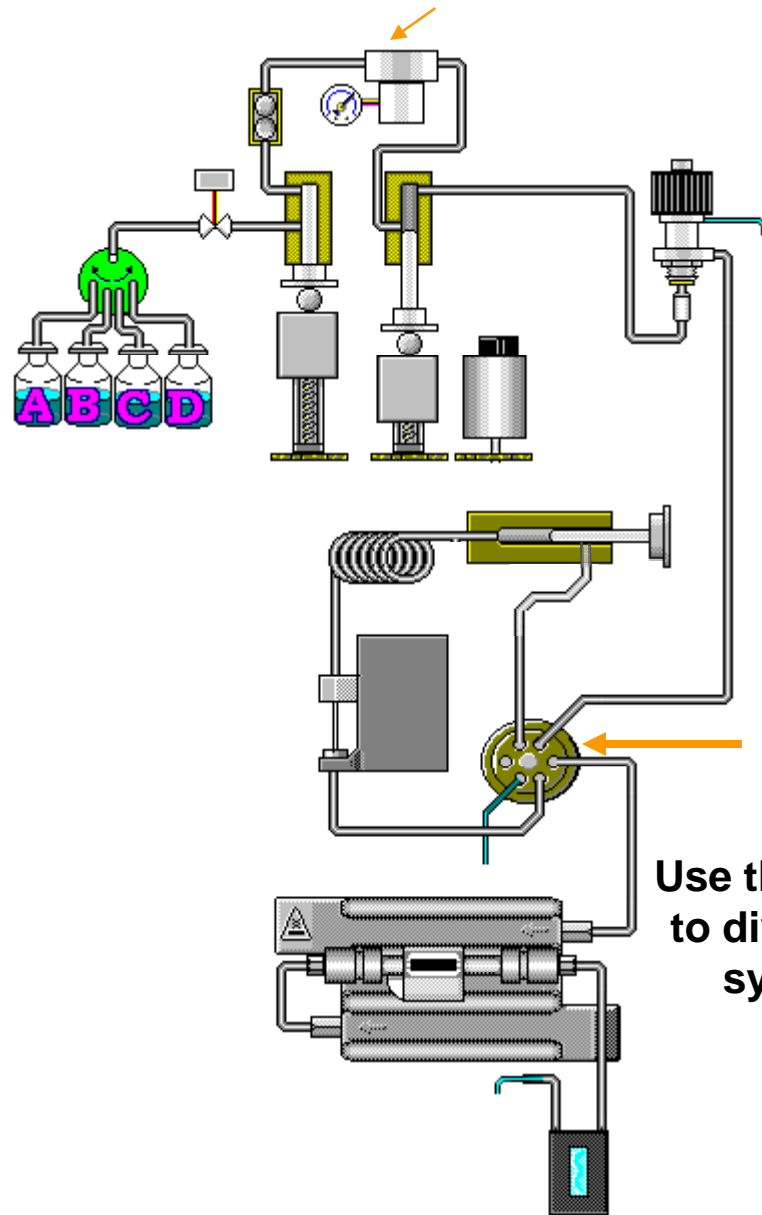


Pressure Problem I

Pressure Too High

- Column inlet frit contaminated
- Frit in purge valve contaminated
- Column contaminated
- Blockage in a capillary, particularly needle seat capillary
- Rotor in injection valve plugged
- Injection needle or needle seat plugged

Pressure Measurement



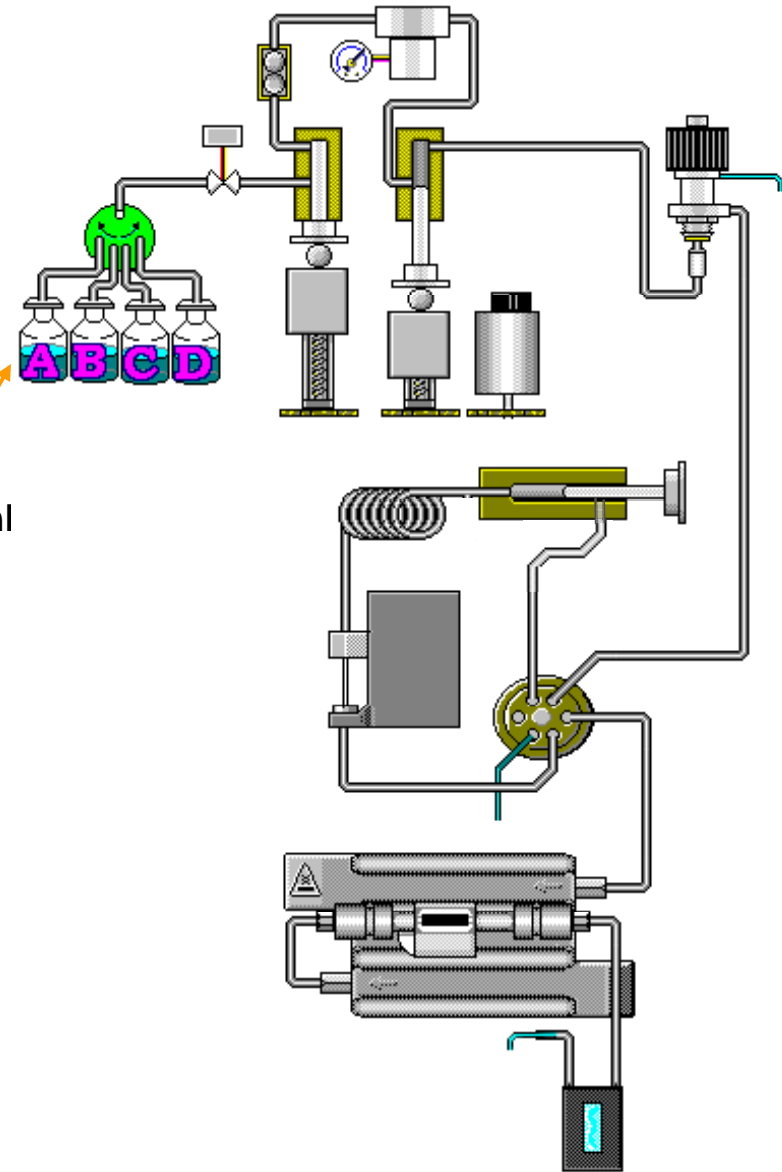
Pressure Problem II

Pressure Too Low

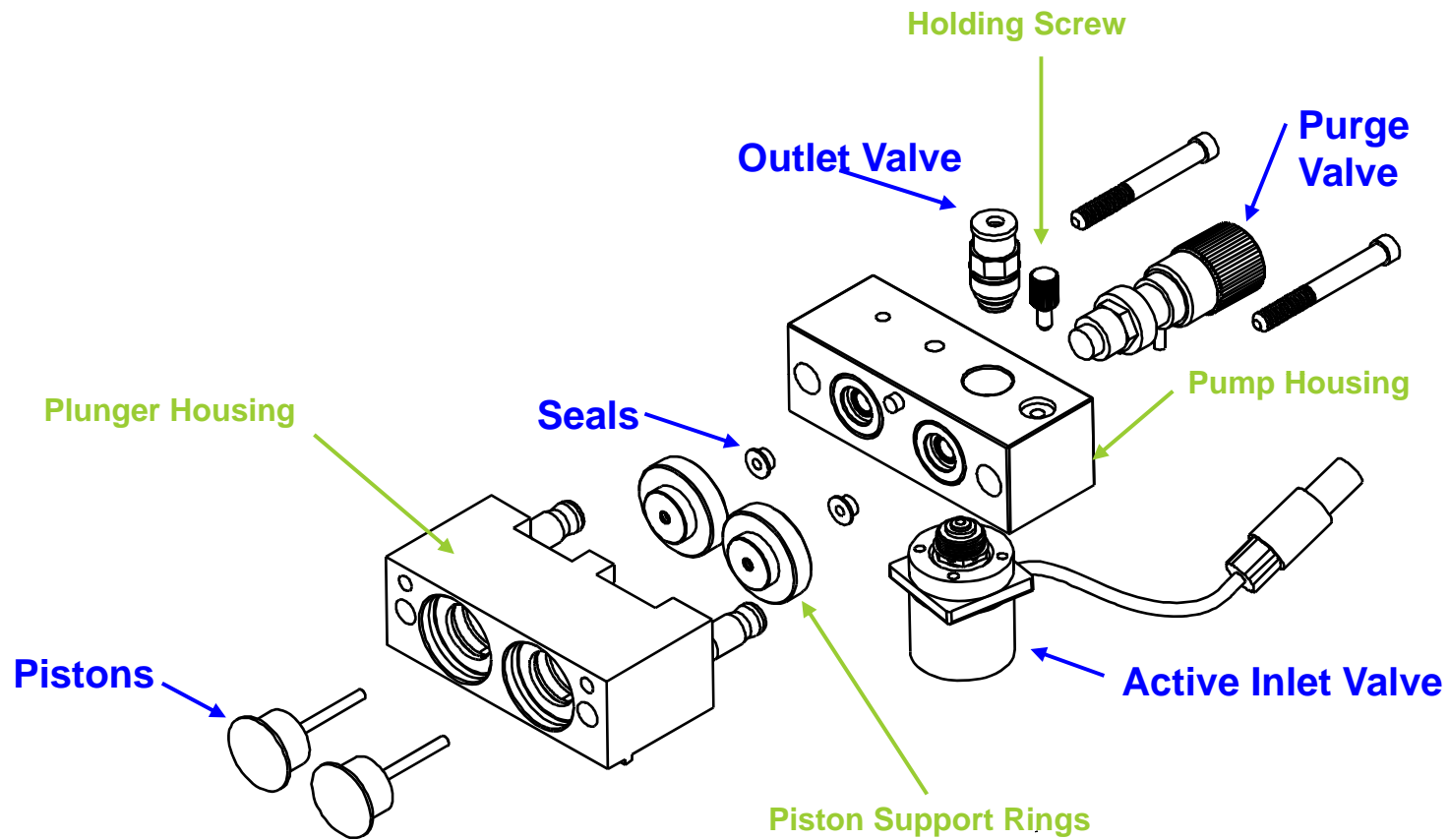
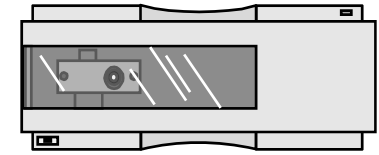
- Solvent inlet frit plugged
- Leak in a capillary connection or other part (pump seals)
- Wrong solvent or flow rate
- AIV (Active inlet valve) defective
- Multichannel Gradient valve incorrectly proportioning
- Ball valve defective
- Column defective (stationary phase)

Solvent inlet frits

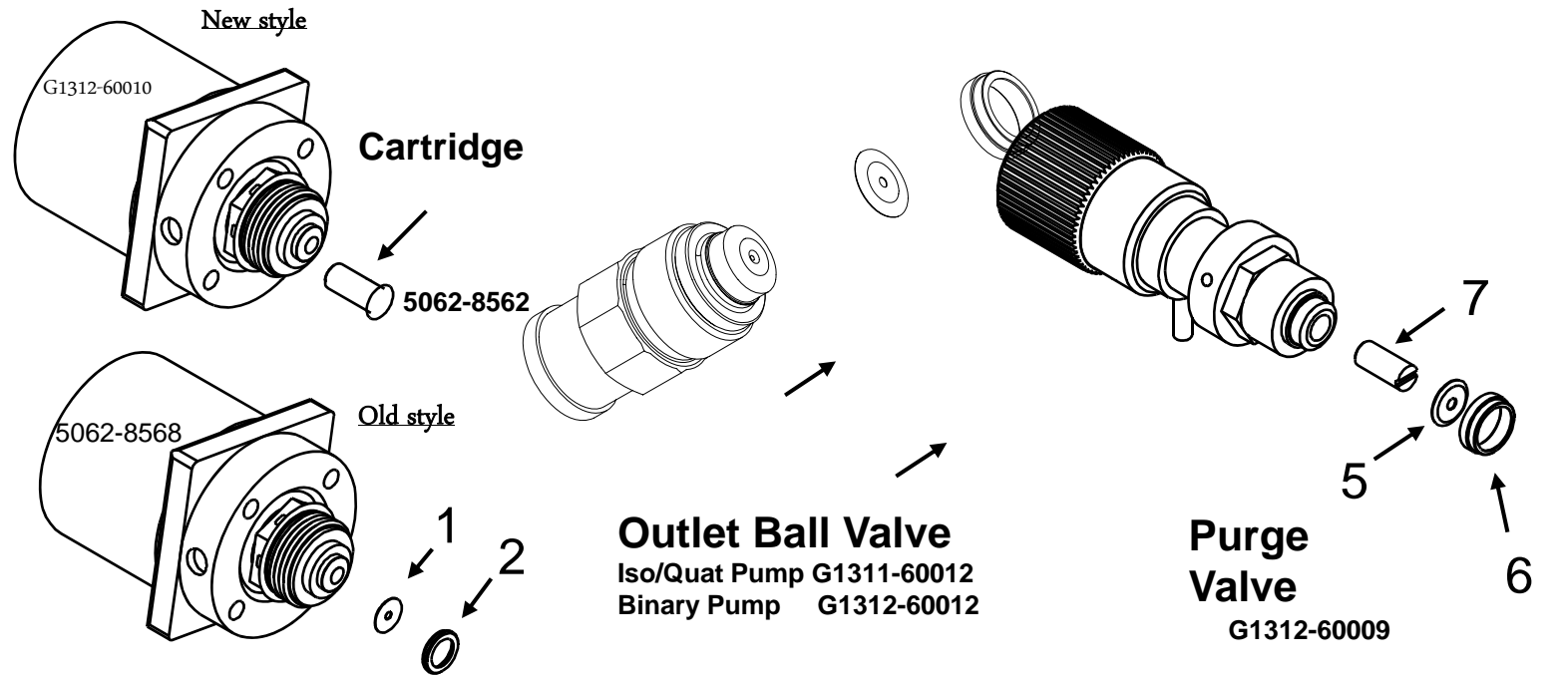
Pressure Measurement



1100 and 1200 Pumps Exploded View



Pump Check Valves



Active Inlet Valve
(common to all)

Outlet Ball Valve
Iso/Quat Pump G1311-60012
Binary Pump G1312-60012

Purge Valve
G1312-60009

- | | | | |
|----------------|-------------|---------------|-------------|
| 1. Gold Washer | 5001-3707 | 5. Gold Seal | 5001-3707 |
| 2. Plastic cap | 01018-21207 | 6. Cap(4pk) | 5062-2485 |
| 3. Gold Seal | 5001-3707 | 7. PTFE (5pk) | 01018-22707 |
| 4. Cap(4pk) | 5062-2485 | | |

Column Cleaning:

**Flush with stronger solvents than your mobile phase.
Make sure detector is taken out of flow path.**

Reversed-Phase Solvent Choices in Order of Increasing Strength

Use at least 10 x V_m of each solvent for analytical columns

1. Mobile phase without buffer salts (water/organic)
2. 100% Organic (MeOH or ACN)
3. Is pressure back in normal range?
4. If not, discard column or consider more drastic conditions:
75% Acetonitrile:25% Isopropanol, then
5. 100% Isopropanol
6. 100% Methylene Chloride*
7. 100% Hexane*

* When using either Hexane or Methylene Chloride the column must be flushed with Isopropanol before returning to your reversed-phase mobile phase.

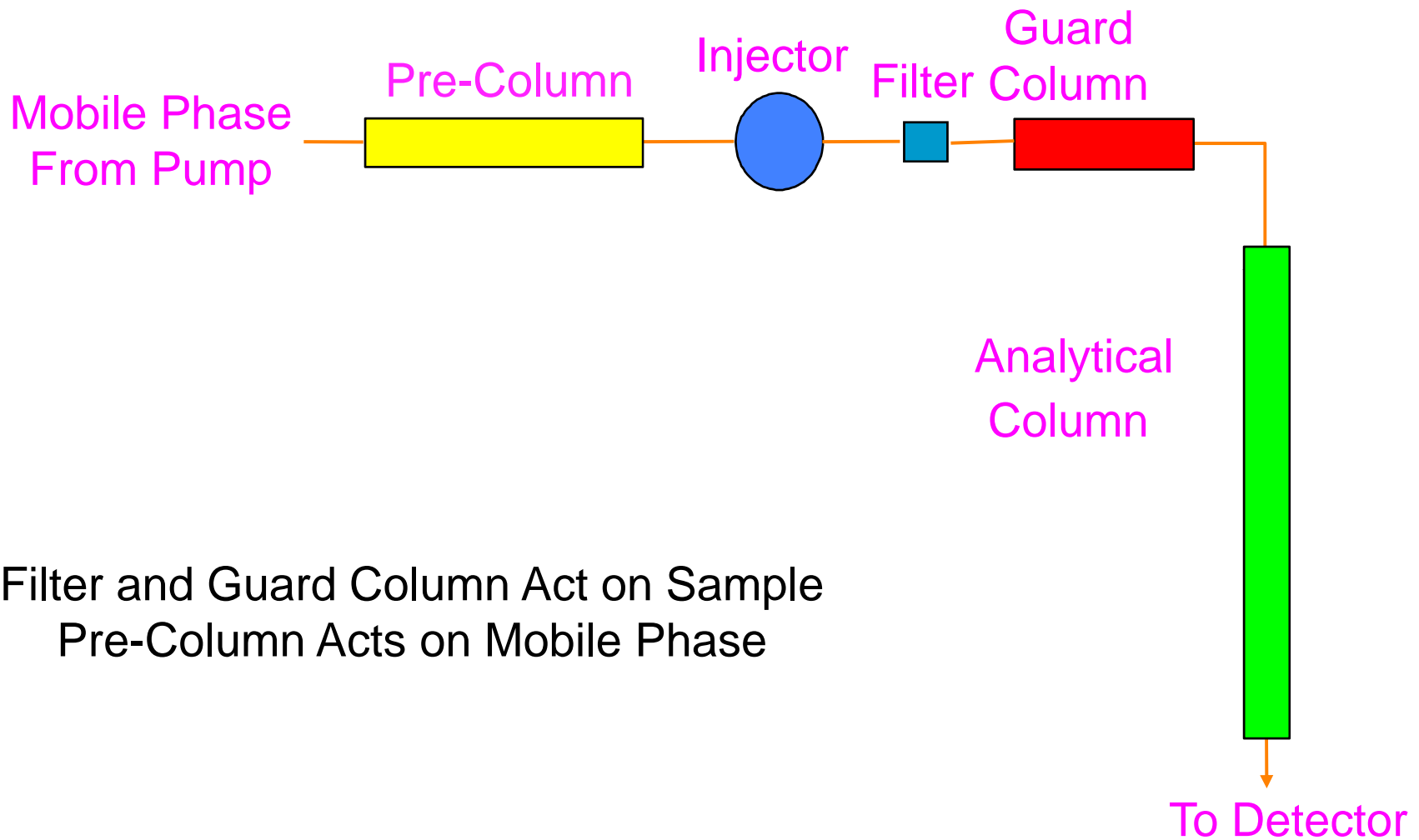
Column Cleaning

Normal Phase Solvent Choices in Order of Increasing Strength

- Use at least 50 mL of each solvent
- 50% Methanol : 50% Chloroform
- 100% Ethyl Acetate

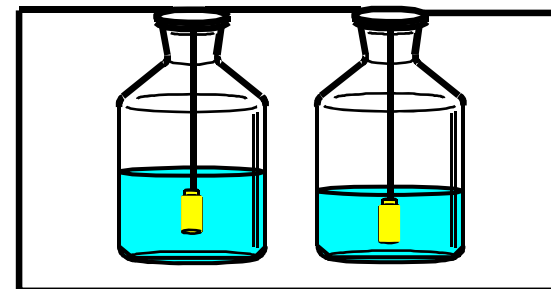


Preventing Back Pressure Problems: In-Line Devices



Filter and Guard Column Act on Sample
Pre-Column Acts on Mobile Phase

Preventing Column Back Pressure Problems:



1. Filter mobile phase:
 - filter non-HPLC grade solvents
 - filter buffer solutions
 - Install an in-line filter between auto-sampler and column (removes pump seal debris, ALS rotor debris, and sample particulates). Use 2 μm frit for 3.5 μm columns, use 0.5 μm frit for 1.8 μm columns.
2. Filter all samples and standards
3. Perform sample clean-up (i.e. SPE, LLE) on dirty samples.
4. Appropriate column flushing – flush buffers from entire system at end of day with water/organic mobile phase.

2. Peak Shape Issues in HPLC

- **Split peaks**
 - **Peak tailing**
 - **Broad peaks**
 - **Poor efficiency (low N)**
 - **Inconsistent Response**
-
- Many peak shape issues are also combinations - i.e. broad and tailing or tailing with increased retention



Split Peaks

Can be caused by:

- Column contamination
- Partially plugged frit
- Column void (gap in packing bed)
- Injection solvent effects

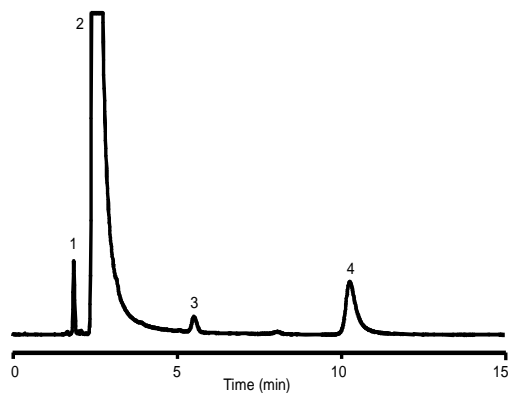


Split Peaks

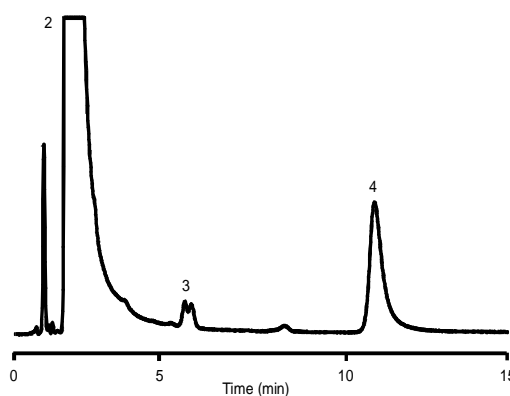
Column Contamination

Column: StableBond SB-C8, 4.6 x 150 mm, 5 μ m Mobile Phase: 60% 25 mM Na₂HPO₄, pH 3.0 : 40% MeOH Flow Rate: 1.0 mL/min
Temperature: 35°C Detection: UV 254 nm Sample: Filtered OTC Cold Medication: 1. Pseudoephedrine 2. APAP 3. Unknown 4. Chlorpheniramine

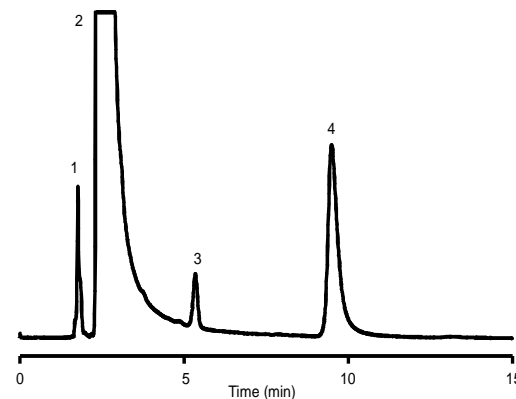
Injection 1



Injection 30



Injection 1
After Column Wash
with 100% ACN



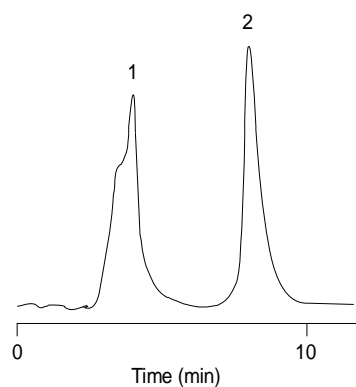
- Column washing eliminates the peak splitting, which resulted from a contaminant on the column.

Split Peaks

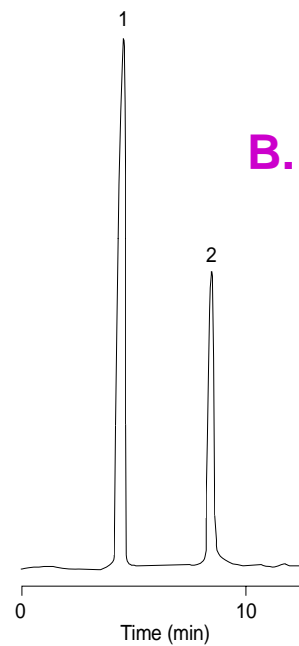
Injection Solvent Effects

Column: StableBond SB-C8, 4.6 x 150 mm, 5 mm Mobile Phase: 82% H₂O : 18% ACN
Injection Volume: 30 mL Sample: 1. Caffeine 2. Salicylamide

**A. Injection Solvent
100% Acetonitrile**

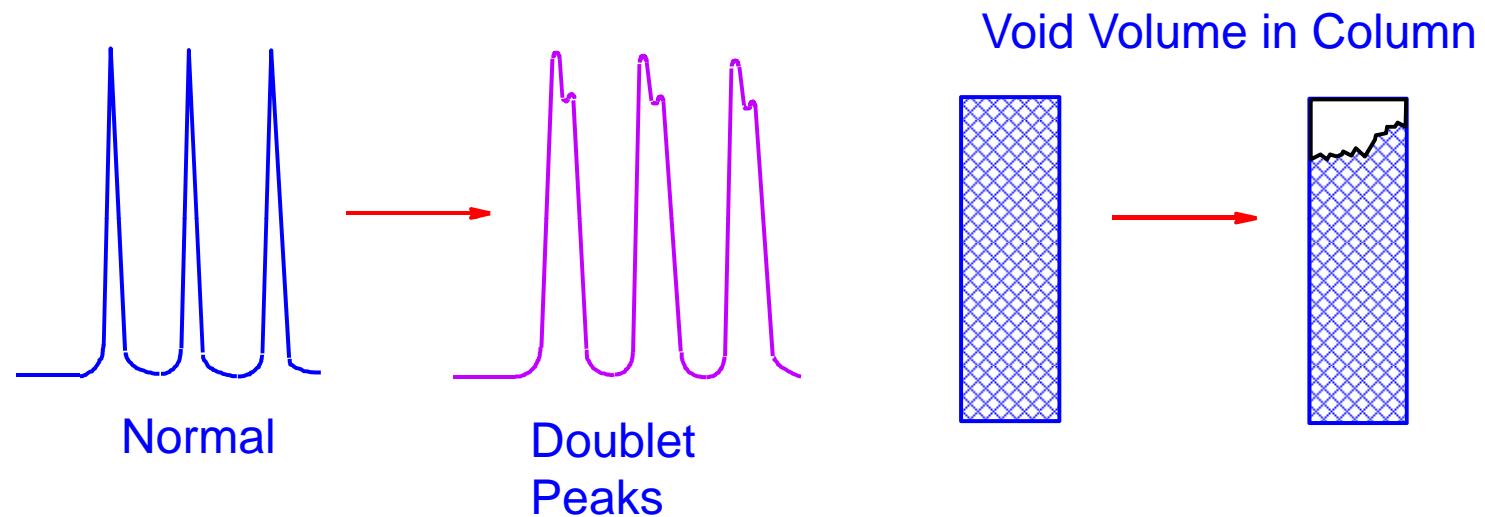


**B. Injection Solvent
Mobile Phase**



- Injecting in a solvent stronger than the mobile phase can cause peak shape problems, such as peak splitting or broadening.
- Note: earlier peaks (low *k*) most affected

Peak Shape Problems - Doublets



- Void Volume in Column
- Partially Blocked Frit
- Only One-Peak a Doublet- Coeluting Components
- Early (low k) peaks most affected

Determining the Cause of Split Peaks

1. Complex sample matrix or many samples analyzed - likely column contamination or partially plugged column frit.
2. Mobile phase pH > 7 - likely column void due to silica dissolution (unless specialty column used, Zorbax Extend-C18 stable to pH 11)
3. Injection solvent stronger than mobile phase - likely split *and* broad peaks, shape dependent on injection volume and k value.

Peak Tailing, Broadening and Loss of Efficiency (N, plates)

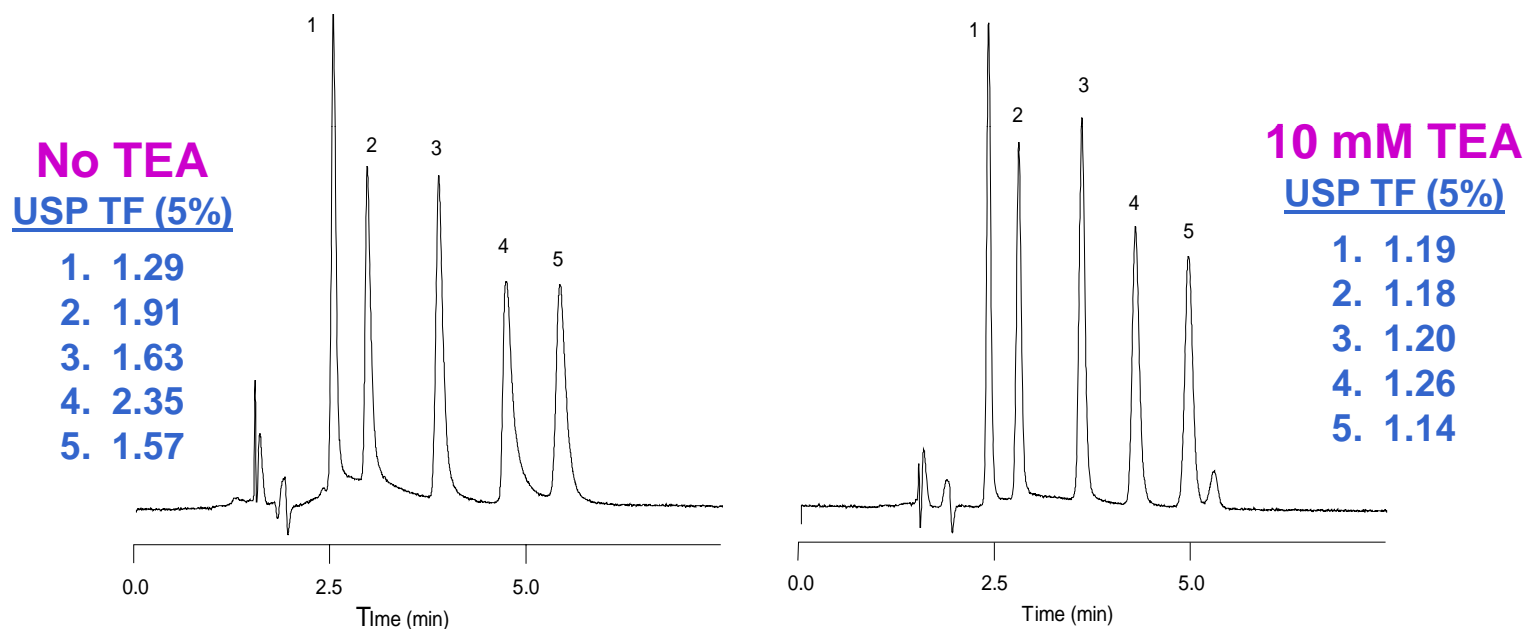
May be caused by:

1. Column “secondary interactions”
2. Column packing voids
3. Column contamination
4. Column aging
5. Column loading
6. Extra-column effects



Peak Tailing Column “Secondary Interactions”

Column: Alkyl-C8, 4.6 x 150 mm, 5mm Mobile Phase: 85% 25 mM Na₂HPO₄ pH 7.0 : 15% ACN Flow Rate: 1.0 mL/min
Temperature: 35°C Sample: 1. Phenylpropanolamine 2. Ephedrine 3. Amphetamine 4. Methamphetamine 5. Phenteramine

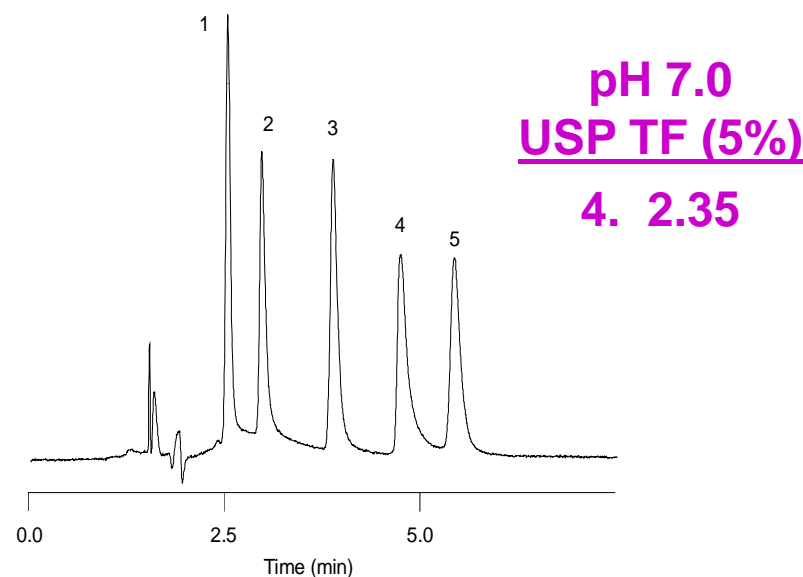
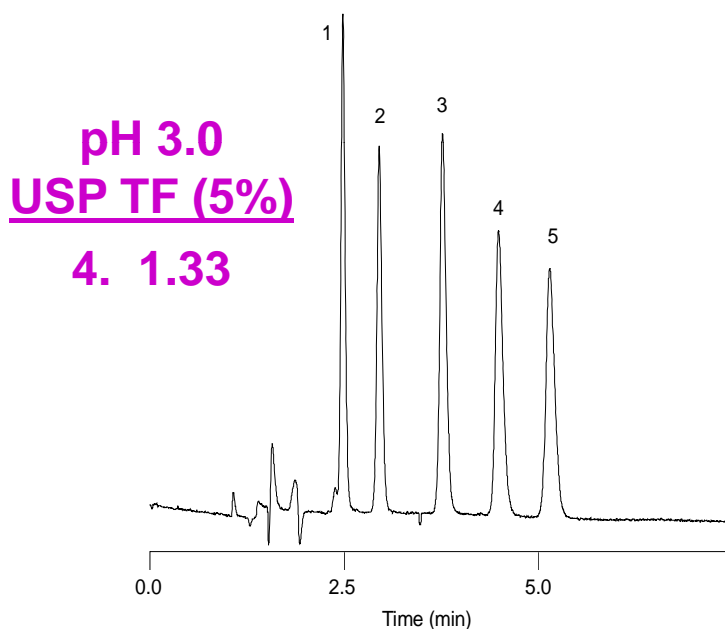


- Peak tailing of amine analytes eliminated with mobile phase modifier (TEA, triethylamine) at pH 7

Peak Tailing

Column “Secondary Interactions”

Column: Alkyl-C8, 4.6 x 150 mm, 5mm Mobile Phase: 85% 25 mM Na₂HPO₄ : 15% ACN Flow Rate: 1.0 mL/min
Temperature: 35°C Sample: 1. Phenylpropanolamine 2. Ephedrine 3. Amphetamine 4. Methamphetamine 5. Phenteramine



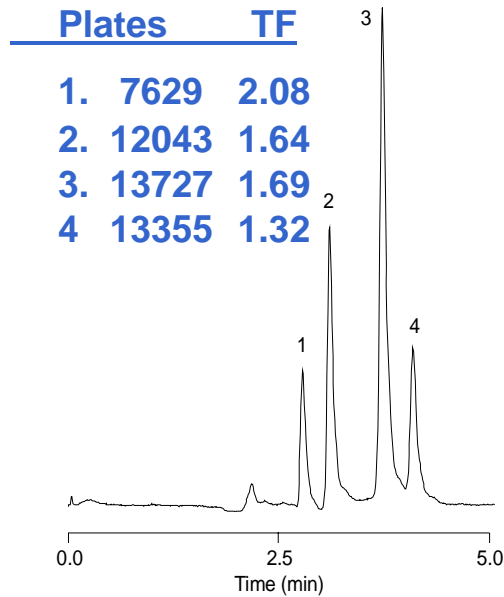
- Reducing the mobile phase pH reduces interactions with silanols that cause peak tailing. No TEA modifier required.

Peak Tailing

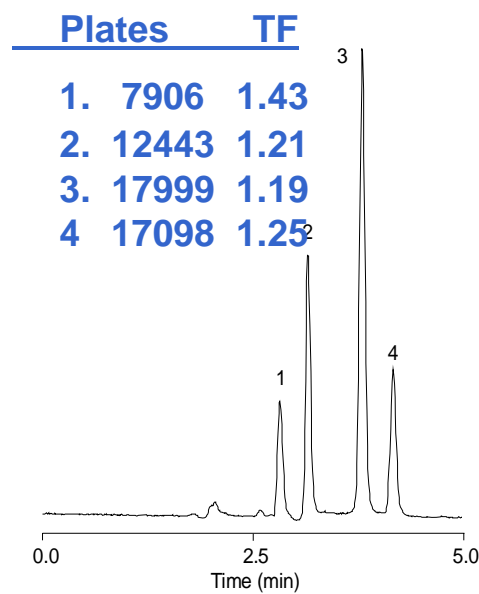
Column Contamination

Column: StableBond SB-C8, 4.6 x 250 mm, 5mm Mobile Phase: 20% H₂O : 80% MeOH Flow Rate: 1.0 mL/min
 Temperature: R.T. Detection: UV 254 nm Sample: 1. Uracil 2. Phenol 3. 4-Chloronitrobenzene 4. Toluene

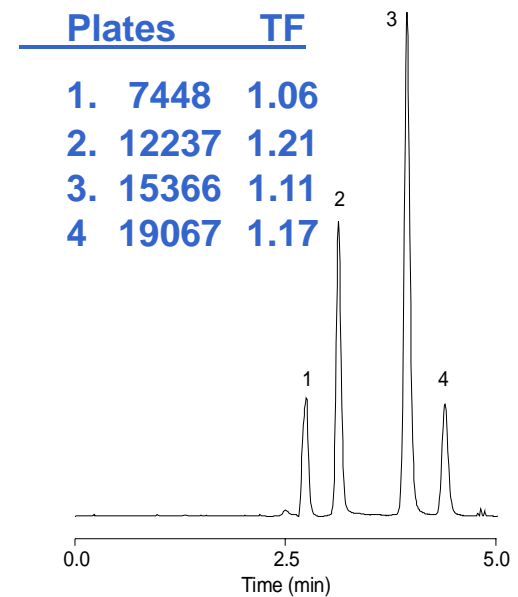
QC test forward direction



QC test reverse direction



QC test after cleaning 100% IPA, 35°C

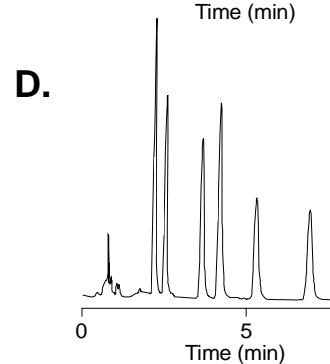
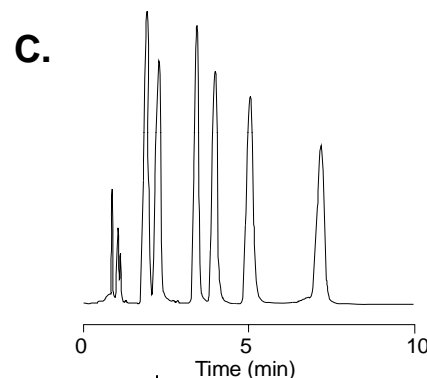
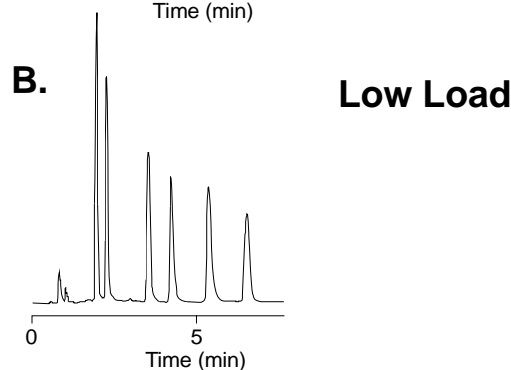
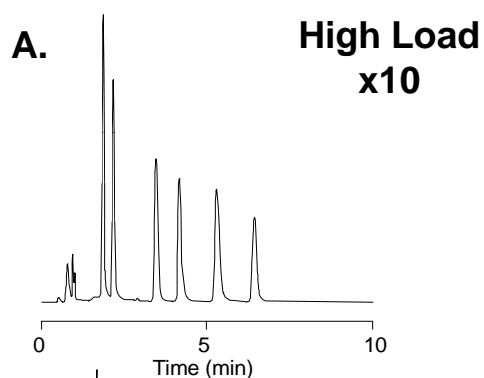


Peak Tailing/Broadening Sample Load Effects

Columns: 4.6 x 150 mm, 5mm Mobile Phase: 40% 25 mM Na₂HPO₄ pH 7.0 : 60% ACN Flow Rate: 1.5 mL/min
 Temperature: 40°C Sample: 1. Desipramine 2. Nortriptyline 3. Doxepin 4. Imipramine 5. Amitriptyline 6. Trimipramine

Tailing
 Eclipse XDB-C8
 USP TF (5%)

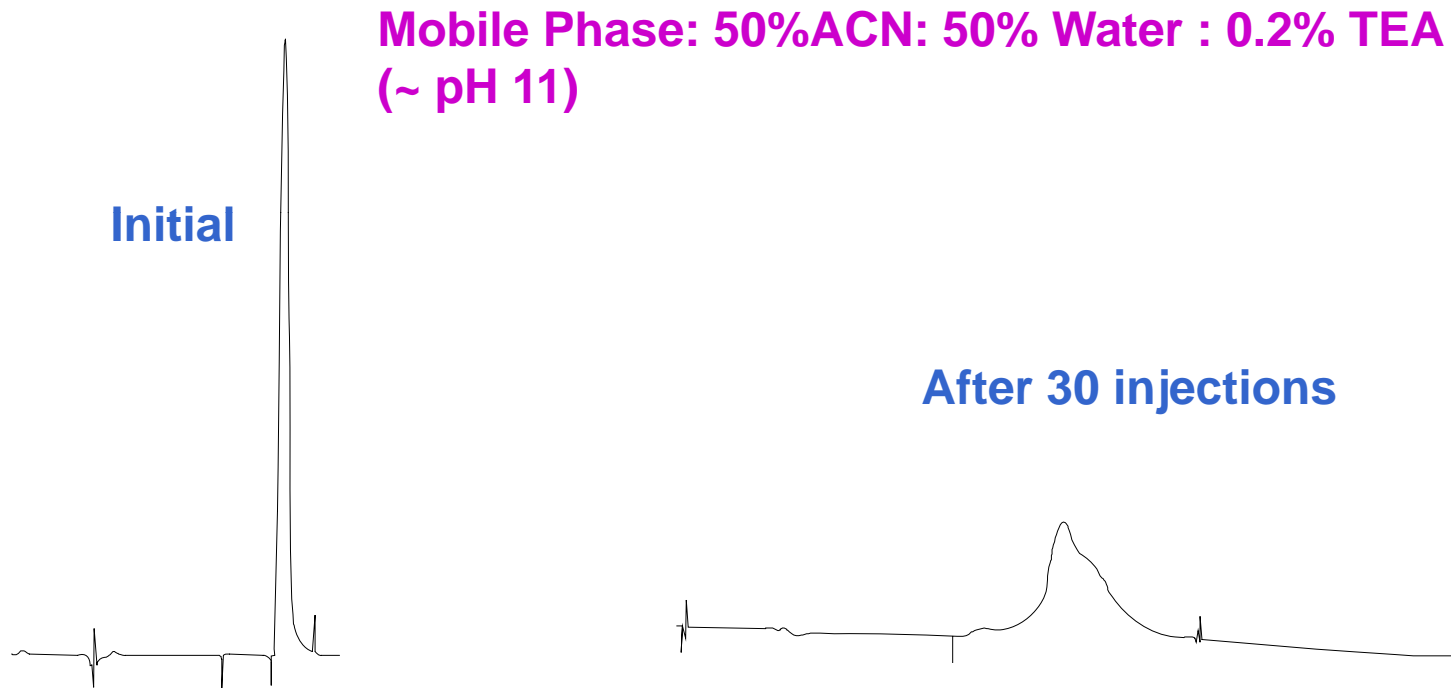
	<u>A</u>	<u>B</u>
1.	1.60	1.70
2.	2.00	1.90
3.	1.56	1.56
4.	2.13	1.70
5.	2.15	1.86
6.	1.25	1.25



Broadening
 Competitive C8
 Plates

	<u>C</u>	<u>D</u>
1.	850	5941
2.	815	7842
3.	2776	6231
4.	2539	8359
5.	2735	10022
6.	5189	10725

Peak Broadening, Splitting Column Void



- Multiple peak shape changes can be caused by the same column problem. In this case a void resulted from silica dissolved at high pH.

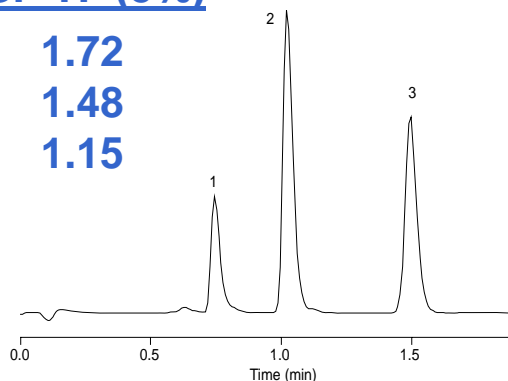
Peak Tailing

Injector Seal Failure

Column: Bonus-RP, 4.6 x 75 mm, 3.5 mm Mobile Phase: 30% H₂O : 70% MeOH Flow Rate: 1.0 mL/min
Temperature: R.T. Detection: UV 254 nm Sample: 1. Uracil 2. Phenol 3. N,N-Dimethylaniline

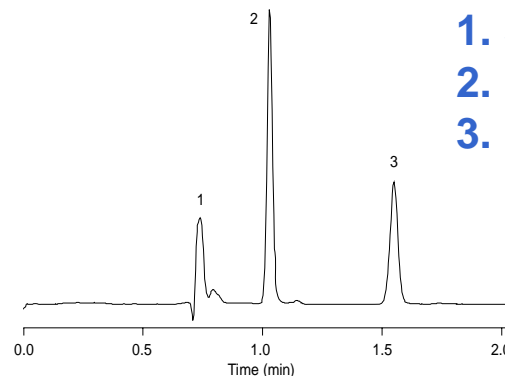
Before

<u>Plates</u>	<u>USP TF (5%)</u>
1. 2235	1.72
2. 3491	1.48
3. 5432	1.15



**After replacing rotor seal
and isolation seal**

<u>Plates</u>	<u>USP TF (5%)</u>
1. 3670	1.45
2. 10457	1.09
3. 10085	1.00

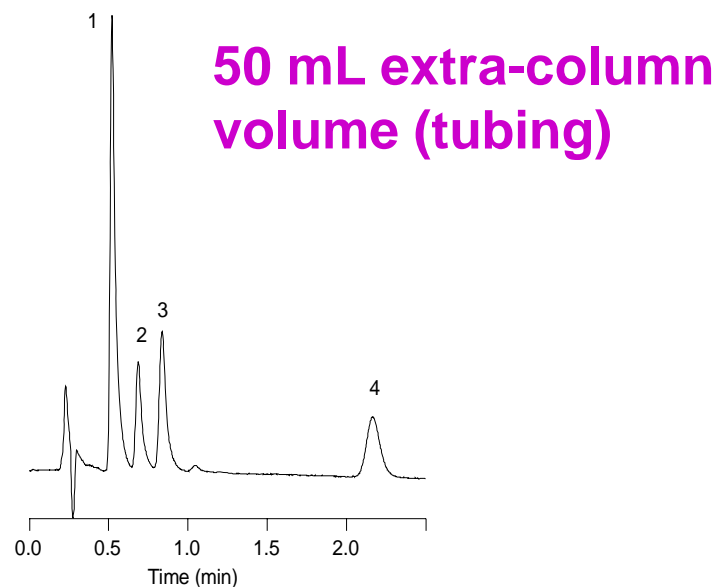
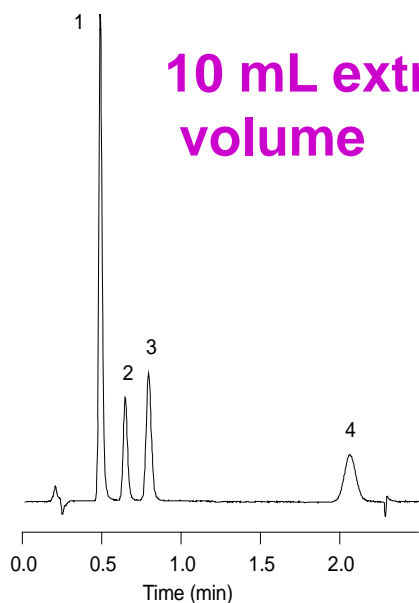


- **Overdue instrument maintenance can sometimes cause peak shape problems.**

Peak Tailing

Extra-Column Volume

Column: StableBond SB-C18, 4.6 x 30 mm, 3.5 mm Mobile Phase: 85% H₂O with 0.1% TFA : 15% ACN Flow Rate: 1.0 mL/min
Temperature: 35°C Sample: 1. Phenylalanine 2. 5-benzyl-3,6-dioxo-2-piperazine acetic acid 3. Asp-phe 4. Aspartame



Determining the Cause of Peak Tailing

- Evaluate mobile phase effects - alter mobile phase pH and additives to eliminate secondary interactions
- Evaluate column choice - try column with high purity silica or different bonding technology
- Reduce sample load – vol inj and concentration
- Eliminate extra-column effects – tubing, fittings, Uv cell
- Flush column and check for aging/void

Reproducibility

Peak retention time precision:

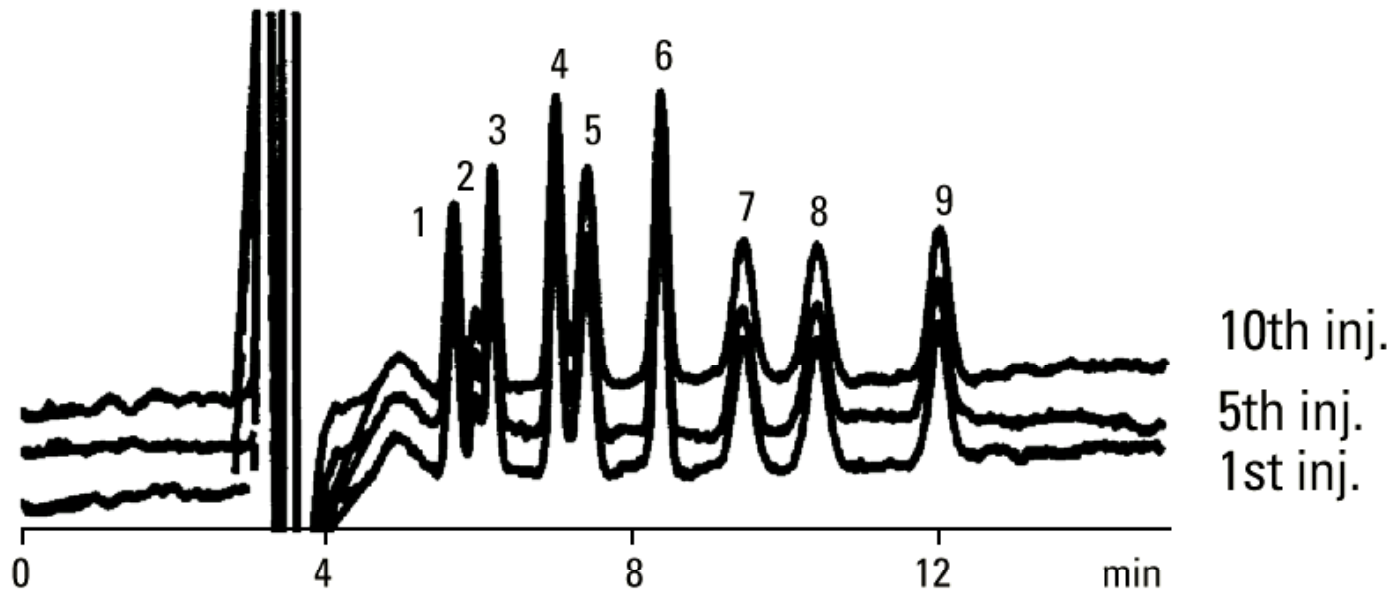
⇒ with oven: _____ < 0.3%

⇒ without oven: _____ < 0.7%

Peak area precision: ≤1.5%

Typically,

- Area and Peak Height problems together point to the autosampler system
- Area and Retention Time problems together point to the pump



Problems with Reproducibility – Peak Areas

Peak Areas not Reproducible

With peak height

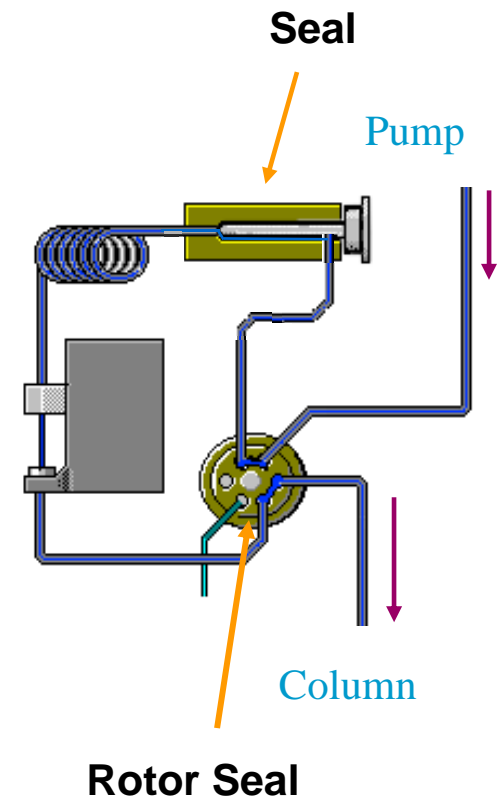
- Rotor seal cross-port leak or injection valve not tight
- Piston seal of metering unit leaking
- Needle partially blocked

With retention time

- Variable pump flow rate

Other

- Capillary from injector to detector not tight
- Detector equilibration problems



3. Retention Issues

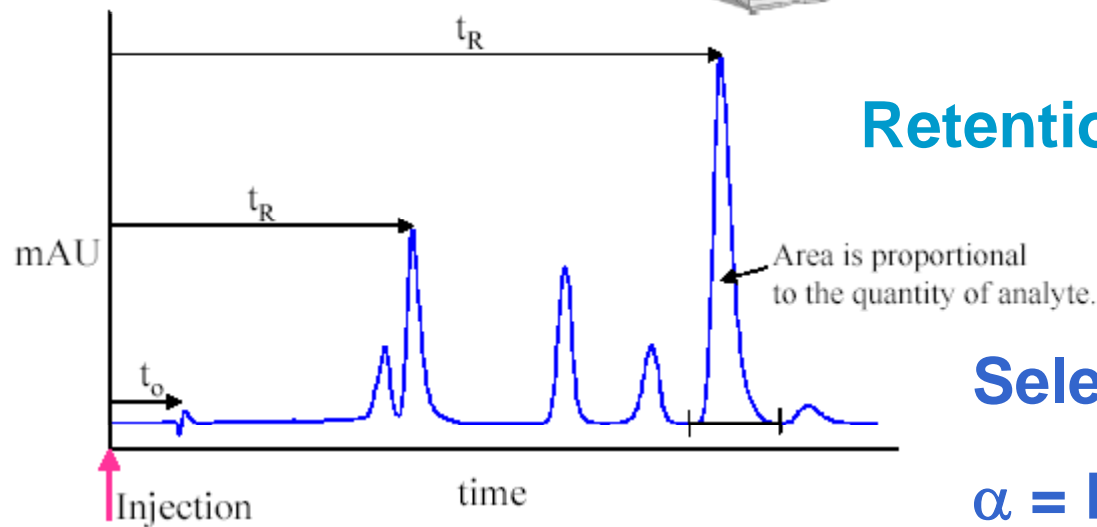
- Retention time changes (t_r)
- Retention factor changes (k')
- Selectivity changes (α)



Retention time t_R , Retention factor k' , and Selectivity factor α

The Chromatogram

t_0 - elution time of unretained peak
 t_R - retention time - determines sample identity



$$\text{Retention factor } k' = (t_R - t_0) / t_0$$

Selectivity factor α

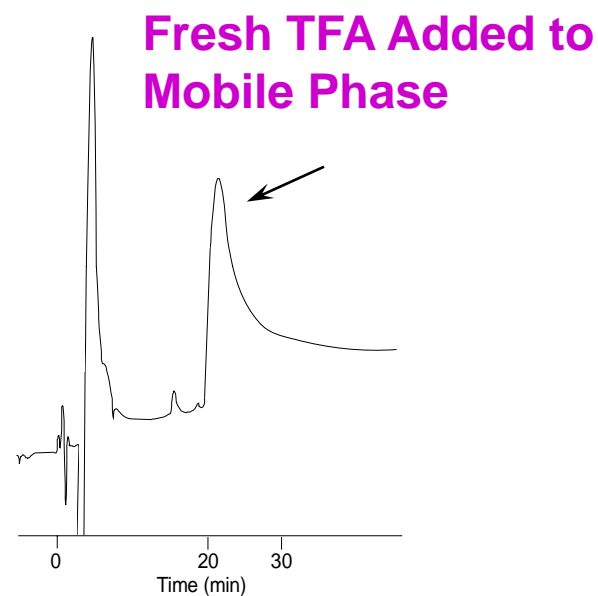
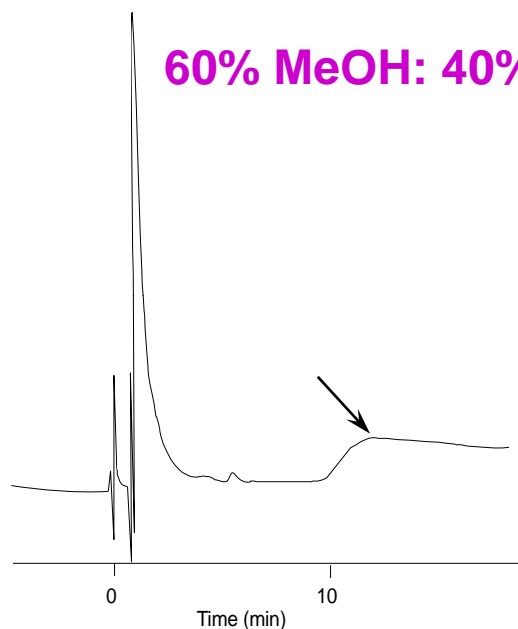
$$\alpha = k_2 / k_1$$

Changes in Retention (k) - Same Column, Over Time

May be caused by:

- 1. Column aging**
- 2. Column contamination**
- 3. Insufficient column equilibration**
- 4. Poor column/mobile phase combination**
- 5. Change in mobile phase**
- 6. Change in flow rate**
- 7. Change in column temperature**
- 8. Other instrument issues**

Mobile Phase Change Causes Change in Retention



- Volatile TFA evaporated/degassed from mobile phase. Replacing it solved problem.
- Chromatography is from a protein binding study and peak shape as expected.

Separation Conditions That Cause Changes in Retention*

Flow Rate	± 1%	± 1% t_r
Temp	± 1° C	± 1 to 2% t_r
%Organic	± 1%	± 5 to 10% t_r
pH	± 0.01%	± 0 to 1% t_r

*excerpted from “Troubleshooting HPLC Systems”, J. W. Dolan and L. R. Snyder, p 442.

Determining the Cause of Retention Changes

Same Column

1. Determine k' , a , and t_r for suspect peaks
2. Wash column
3. Test new column - note lot number
4. Review column equilibration procedures
5. Make up fresh mobile phase and test
6. Check instrument performance

Change in Retention/Selectivity

Column-to-Column

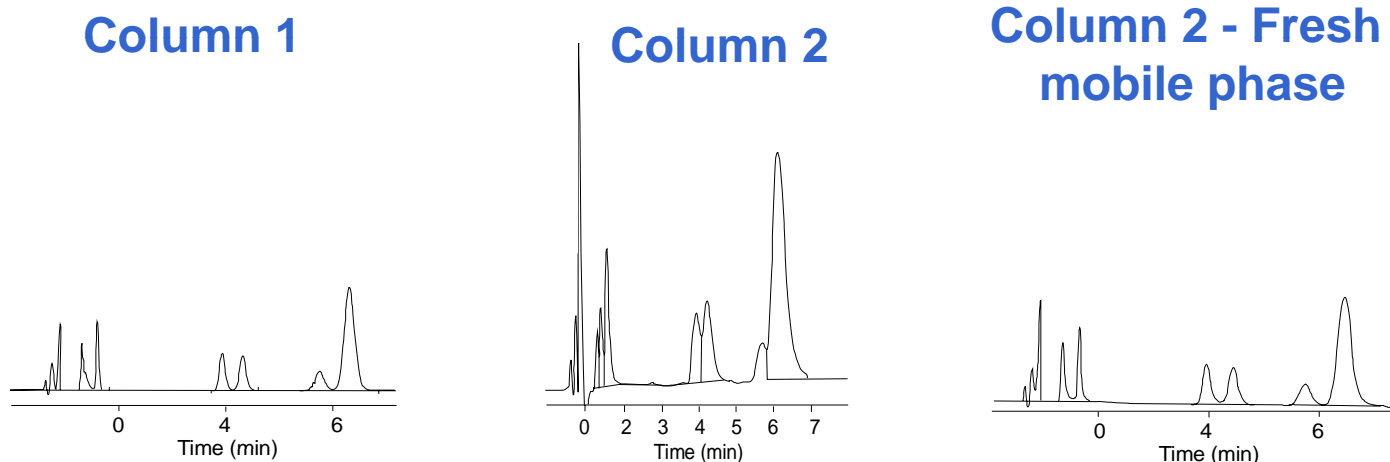
1. Different column histories (aging)
2. Insufficient/inconsistent equilibration
3. Poor column/mobile phase combination
4. Change in mobile phase
5. Change in flow rate
6. Other instrument issues
7. Slight changes in column bed volume (t_r only)



Example Change in Retention/Selectivity

Column-to-Column

Mobile Phase Variation



“I have experimented with our mobile phase, opening new bottles of all mobile phase components. When I use all fresh ingredients, the problem ceases to exist, and I have narrowed the problem to either a bad bottle of TEA or phosphoric acid. Our problem has been solved.”

Minimize Change in Retention/Selectivity

Lot-to-Lot

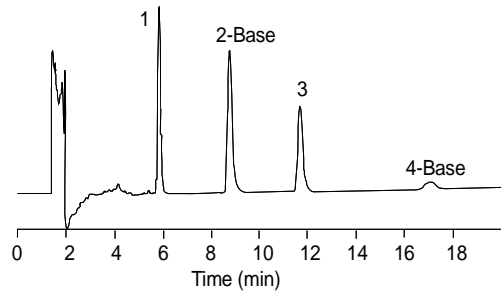
Evaluate:

1. All causes of column-to-column change*
2. Method ruggedness (buffers/ionic strength)
3. pH sensitivity (sample/column interactions)

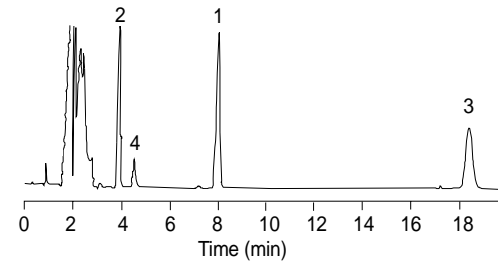
*All causes of column-to-column change should be considered first, especially when only one column from a lot has been tested.

Lot-to-Lot Selectivity Change - pH

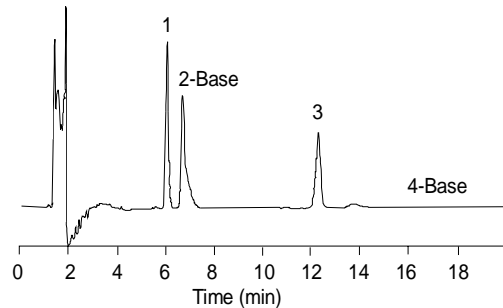
pH 4.5 - Lot 1



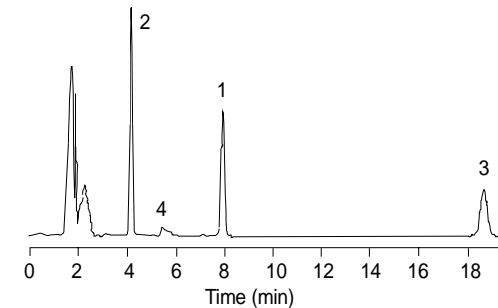
pH 3.0 - Lot 1



pH 4.5 - Lot 2



pH 3.0 - Lot 2



- pH 4.5 shows selectivity change from lot-to-lot for basic compounds
- pH 3.0 shows no selectivity change from lot-to-lot, indicating silanol sensitivity at pH 4.5
- Evaluate several pH levels to establish most robust choice of pH

Problems with Reproducibility – Peak Areas

Peak Areas not Reproducible

With peak height

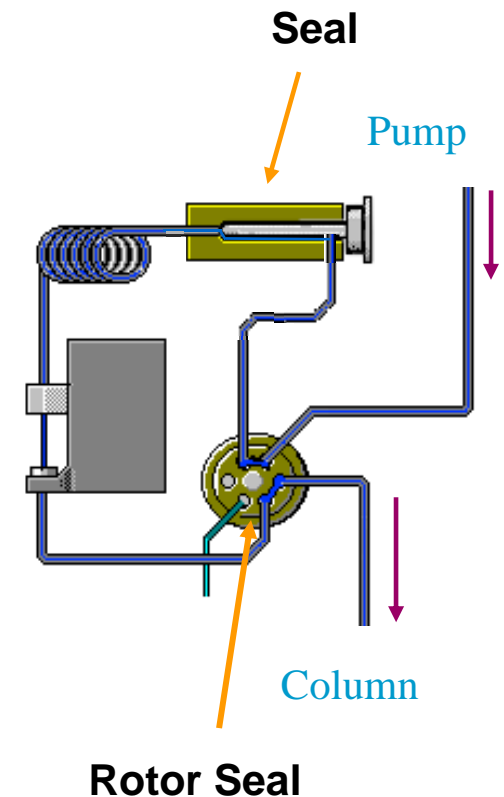
- Rotor seal cross-port leak or injection valve not tight
- Piston seal of metering unit leaking
- Needle partially blocked

With retention time

- Variable pump flow rate

Other

- Capillary from injector to detector not tight
- Detector equilibration problems



Problems with Reproducibility – Retention Time

Retention Times not Reproducible

- **Pump Problems**

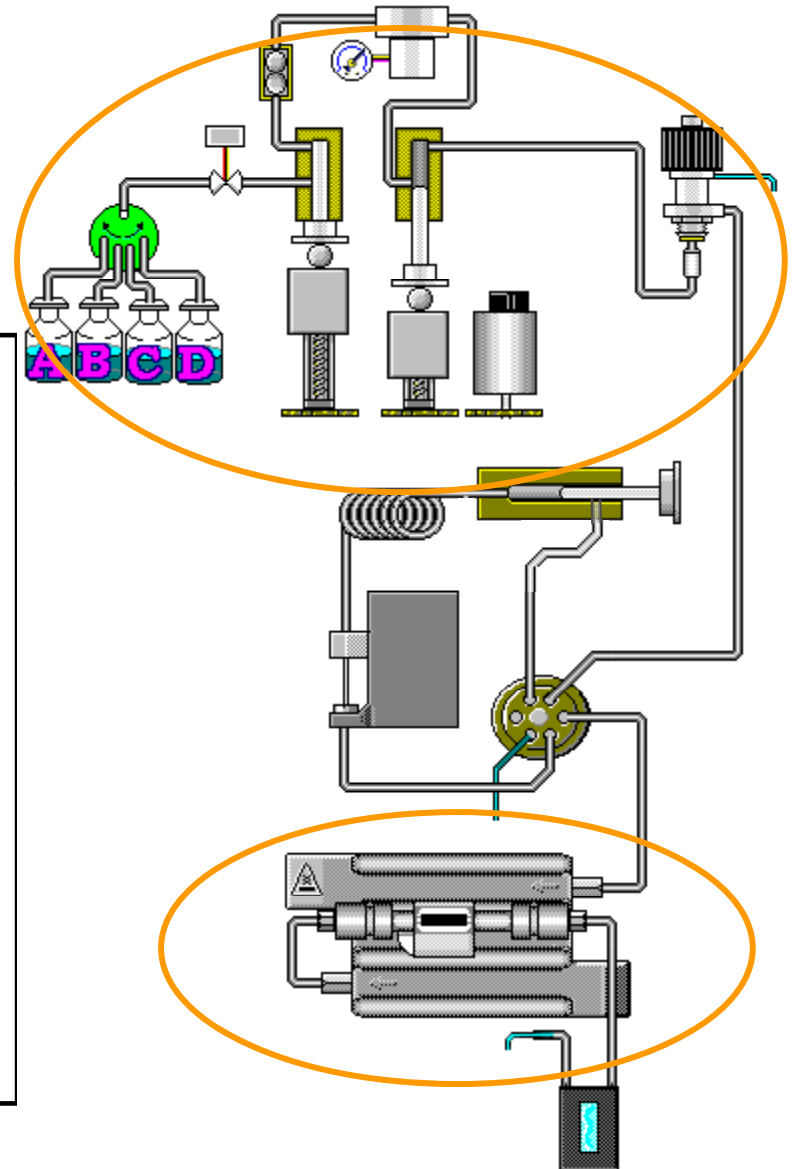
- Mobile phase composition problems
- Valves AIV, ball valve defective
- Flow rate problems

- **Column Oven Problems**

- Temperature fluctuations

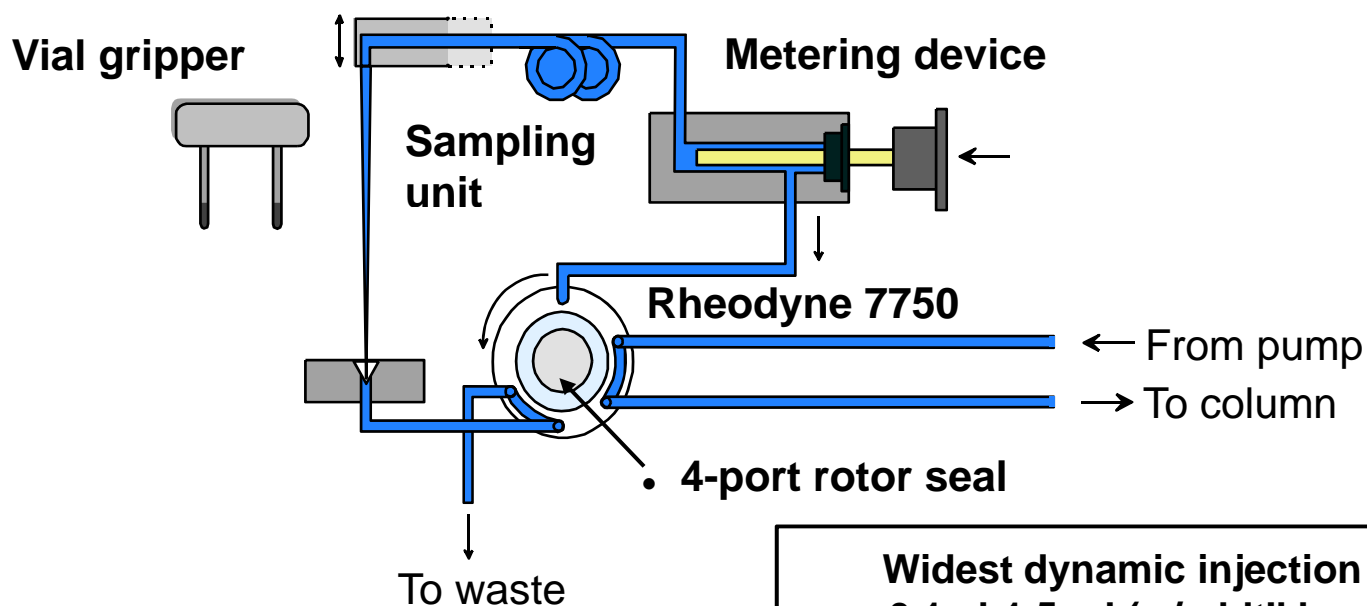
- **Other**

- Column equilibration
- Column deterioration



Autosampler Principle of Operation

Standard loop volume 300 μ l
Total delay volume 300 μ l + V_{inj}
Minimal (bypass) delay volume 6.2 μ l



Widest dynamic injection range:
0.1 μ l-1.5 ml (w/addt'l hardware)

Evaluate Retention Changes

Lot-to-Lot

1. Eliminate causes of column-to-column selectivity change
2. Re-evaluate method ruggedness - modify method
3. Determine pH sensitivity - modify method
4. Classify selectivity changes
5. Contact manufacturer for assistance*

Agilent Column Support: 800-227-9770, option 4, option 2 (LC columns)



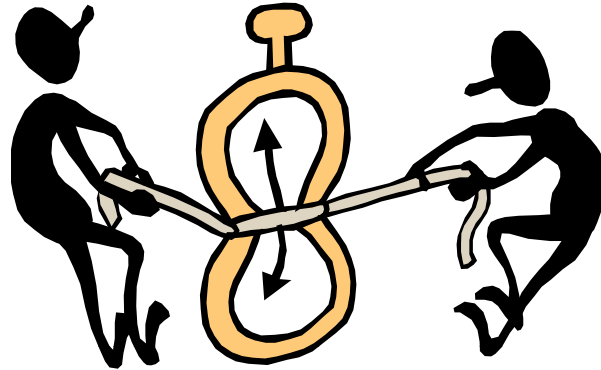
Conclusions:

HPLC column problems are evident as:

1. High pressure
2. Undesirable peak shape
3. Changes in retention/selectivity

These problems are not always associated with the column and may be caused by instrument and experimental condition issues.





The End – Thank You!

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